

RESEARCH ARTICLE

Antibacterial Potency of Ozonated Water against *Escherichia coli*

Syarifa Fitria¹, Zolkafle Buntat², Zainuddin Nawawi^{3*}, Muhammad Abu Bakar Sidik³, M.I. Jambak³ and Dwirina Yuniarti³

¹Graduate School of Environmental Science, University of Sriwijaya, Palembang, South Sumatra 30128, Indonesia.

²Institute of High Voltage and High Current, Faculty of Electrical Engineering, Universiti Teknologi Malaysia, Johor Baru 81310, Malaysia. ³Department of Electrical Engineering, Faculty of Engineering, University of Sriwijaya, Palembang, South Sumatra 30128, Indonesia.

Abstract

Sterilization is essential for inactivation of microorganisms. There are many methods of sterilization, such as the use of heat or chemical processes. However, some equipment can be damaged by heat and can only be sterilized at low temperatures. Failure to properly disinfect or sterilize equipment may lead to transmission via contaminated objects. This paper presents a sterilization process using ozonated water at a temperature of 29.5°C with gram-negative bacteria (*Escherichia coli*). The antibacterial effect was examined with various concentrations of ORP (oxidation reduction potential) at 702 mV, 802 mV, 940 mV, 950 mV, and 960 mV. A strong linear correlation was observed between ORP value and the surface area of the antibacterial effect. It was found that increasing the concentration of ORP affects the surface area of *Escherichia coli*.

Keywords: Antibacterial, *E. coli*, ozonated water, oxidation reduction potential.

*Correspondence: nawawi_z@yahoo.com+62-711-580069

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INTRODUCTION

Sterilization is important for inactivation of microorganisms, providing improved good quality of life for humans. Inactivation of a microorganism includes its destruction or elimination by physical or chemical processes or a combination of the both¹. Application of conventional sterilization methods, such as using high temperature, high pressure, chemical gas, and radiation (gamma rays), depends on the type of materials being sterilized.

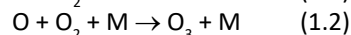
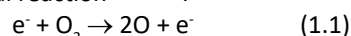
Although some methods using high temperature and high pressure are effective in inactivating microorganisms, they are inefficient regarding sterilization duration, energy consumption, and plastic equipment application^{1,2,3}. Sterilization using chemicals, gas, and radiation can be applied to many types of equipment. However, the method will produce toxic residues, change molecular structure (cross-link or scissor), release odors, change pH, cause discoloration and degradation of a few materials, or affect bond strengths and change over the shelf life of the material³.

Another way to sterilize is through the use of ozone as a disinfectant. Use of ozone and ionized water is an environmentally friendly method for inactivation of bacterial^{4,5}. Ozone is a powerful and effective germicidal oxidant that has great potential over chlorine (chemical method) and other disinfectant methods. Ozone is currently used as a disinfectant for water, air, and various pharmaceutical applications^{6,7,8}. The most common method for ozone generation is Dielectric Barrier Discharge (DBD). The DBD model is shown in Fig. 1. This method consists of at least one insulating layer between two electrodes or cylindrical electrodes that connect to an AC power supply with a dielectric layer.

Homogeneous discharges produced in the air gap between the electrodes—the volume of the reaction chamber—cause the temperature in the chamber to remain low (25°C), which reduces the need for a cooling system^{9,10,11}. Dry air or oxygen that is supplied in the DBD chamber allows the dissociation of oxygen molecules to form ozone. In 2001, ozone in the gaseous and aqueous phases was accepted by the U.S. Food and Drug Administration (US FDA) as an antimicrobial

agent for the treatment, storage, and processing of foods¹².

Ozone has decomposition products which can rapidly inactivate microorganisms (e.g., hydroxyl radical) by reacting with intracellular enzymes, nucleic material, and components of the cell envelope¹². Inactivation of bacteria using ozone causes leakage of inner contents due to oxidation of unsaturated lipids in the cell envelope, which finally results in cell lysis^{12,13,14}. The mechanism most referred to in the formation of ozone in electrical discharge is the following chemical reaction^{15,16,17,18}.



M is a third component necessary to support the reaction when the air is injected. M could be gas molecules such as oxygen, helium, or argon.

In the reaction, electron bombardment breaks oxygen molecules apart; electrons that avalanche will recombine with each other or with the other oxygen molecules for ozone formation¹⁵. The method used in this experiment aimed to detect the extent that dissolved ozone mixed with distilled water was monitored through the ORP meter. ORP is well-known to have higher efficiency in inactivating bacteria with higher values¹⁹.

Failure to properly disinfect or sterilize equipment may lead to transmission via contaminated objects. The objective of this study was to observe whether the ozonized water with different concentrations of ORP could develop an inhibition zone for *Escherichia coli*. The linear

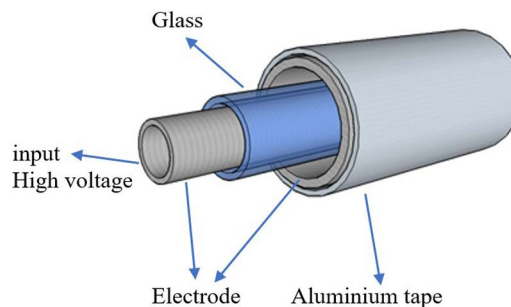


Fig. 1. Dielectric Barrier Discharge model. [Change aluminum to aluminium. Change to High voltage input]

correlation between surface area and rate of oxidation-reduction potential used in ozonized water was examined.

MATERIALS AND METHODS

Escherichia coli ATC9222 was used as a sample for indicating contaminated objects. At first, the *Escherichia coli* was developed in Nutrient Agar slant and then transferred to a nutrient broth composed of 0.65 g and 50 ml of distilled water, and incubated for 24 hours at 35°C. Samples in the nutrient broth that were grown for 24 hours were spread into a petri dish (Anumbra, 100 x 15 mm²). Whatman papers were placed into the petri dish that already dyes into the ozonized water. The results were obtained after the petri dish incubated for 24 hours at 35°C.

Ozone was generated using a dielectric barrier discharge system with an applied voltage of 15 kV at atmospheric pressure. Oxygen with a purity of 99.9% was injected into the ozone generator at a constant flow rate of 0.5 l/min. Electron avalanches processed in the electric field led to the creation of partially ionized plasma and created ozone. Ozone was mixed with distilled water by using a Venturi Injector and a static mixer. From the static mixer, the ozonated water was collected in a bottleneck tube. The ozonated water was measured by an ORP meter (AZ Instrument 8551) and a pH meter (Hanna HI 98107) to analyze the correlation of ORP and pH with the surface area containing *Escherichia coli* in the petri dish (Fig. 2).

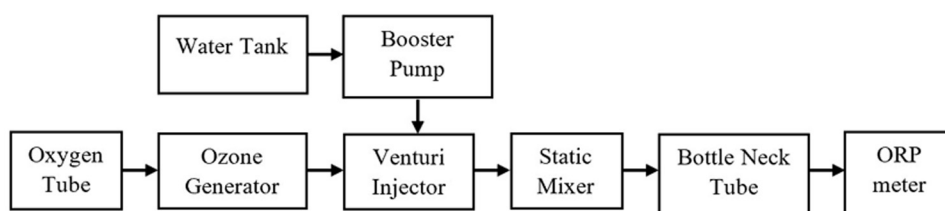


Fig. 2. Experimental setup

The ozonated water in various concentrations of oxidation-reduction potential (702 mV, 802 mV, 940 mV, 950 mV, and 960 mV) was injected into Whatman papers and then put into a petri dish already containing *Escherichia coli* in EMB (eosin methylene blue) media. Formed inhibition zones were obtained after the petri

dish was incubated for 24 hours at 35°C. The circular transparent zones in the exposed samples represent the growth of the inhibition zone. They could be observed directly without special equipment and the diameter measured by using a ruler (1 mm precision).

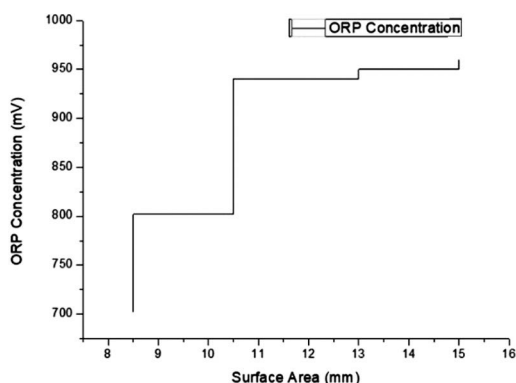


Fig. 3. ORP concentration vs surface area

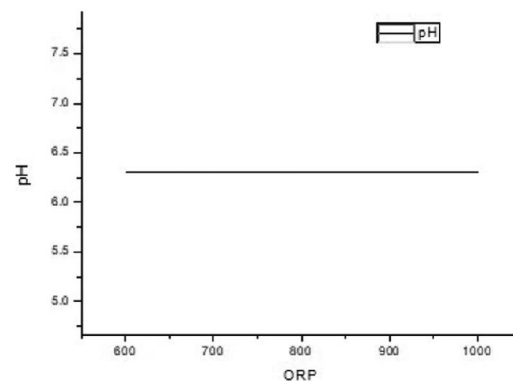


Fig. 4. pH vs ORP concentration

RESULTS AND DISCUSSION

The addition of ozone to the distilled water gives a relatively linear result of changes in the surface area with the different oxidation-reduction potential concentrations used. Fig. 3 provides information on the antibacterial effects on *Escherichia coli* tested with ORP.

The width of the surface area in the *Escherichia coli* stained with EMB shows a beneficial contribution of ozonated water to bacterial growth elimination. The observation of an inhibition zone against bacterial growth was carried out in various concentration over 3 days. At a low concentration of 702 mV, as shown in Fig. 3, the inhibition zone was limited only to the surface area, even in 3 days of observation. As shown in Fig. 4, when using a concentration of 960 mV, the inhibition zone was more significant than in lower concentrations, and no modification was seen in 3 days. This proves that high concentrations of ozonated water can create a beneficial inhibition zone against *Escherichia coli*, as shown in Fig. 5.

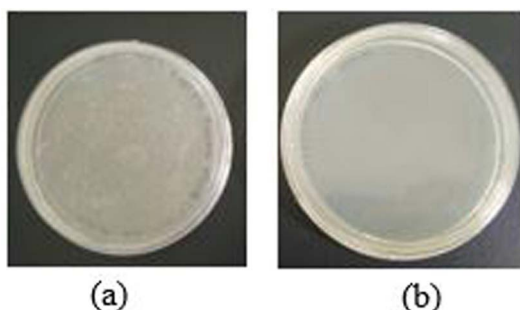


Fig. 5. Inhibition zones against *Escherichia coli*

The ORP value gives the potential redox level of the ozonated water, which has an oxidation value that acts to inhibit and inactivate bacteria. During the inhibition process, the ORP value indicates the oxidative agents whereas pH remained stable. This result agrees with the findings of Wang *et al.*²⁰, oxidation was not influenced by low pH, suggesting that oxidation plays an important role during the process. Based on Fig. 6a and Fig. 6b, ozonated water is able to inactivate *Escherichia coli*, there was no growth after incubation for 24 hours at 35°C. This result has a good agreement with evidence that ozone is known to have antibacterial activity^{21,22} (Fig. 5).

Additional distilled water creates an electron impact reaction in the impulse stage,

also generating H and OH radicals that are able to inactivate bacteria¹⁶. The ORP was considered to be a more suitable indicator for the optimal operation of the antibacterial process (Fig. 6a and Fig. 6b).

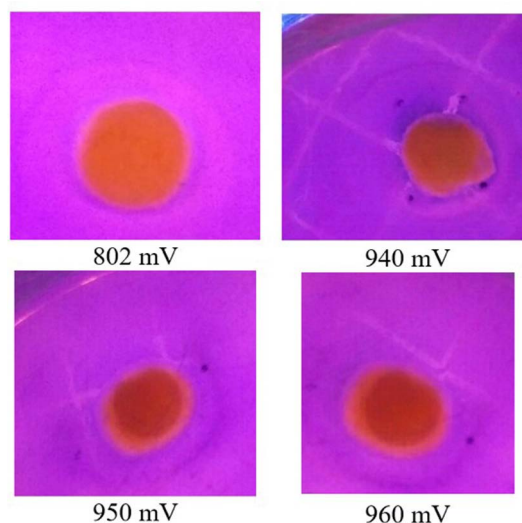


Fig. 6. a. Before treatment; b. After treatment

CONCLUSION

Ozone is known as an antibacterial agent. Ozone mixed with distilled water, that is, ozonated water, has good results for increasing the inhibition zone against *Escherichia coli*. The ORP concentration in ozonated water plays an important role as an oxidative agent for inhibition and inactivation of bacteria but the pH value was almost stable in the entire process. Since, the electron that oxidized in water could affect bacteria growth around surface area.

Further research is needed on the oxidation-reduction potential for other bacteria to determine the inhibition zone at various concentrations and subsequent cell damage after exposure to ozonated water. The proposed system may find future use as a means for sterilizing medical equipment using ozone generated and mixed with distilled water.

ACKNOWLEDGEMENTS

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COFLICT OF INTEREST

The author declares that there is no conflict of interest.

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