

## Fw: Photosynthetica - Manuscript ID 1145-03-2015

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-----Original Message-----

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Sent: Saturday, March 14, 2015 3:48 PM

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Subject: Photosynthetica - Manuscript ID 1145-03-2015

14-Mar-2015

Dear Prof. Nose:

Your manuscript entitled "An improved method for the simultaneous determination of photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption of *Rhizopora mucronata* leaves under aqueous condition" has been successfully submitted online and is presently being given full consideration for publication in *Photosynthetica*.

Your manuscript ID is 1145-03-2015.

Please mention the above manuscript ID in all future correspondence or when calling the office for questions. If there are any changes in your street address or e-mail address, please log in to ScholarOne Manuscripts at <https://mc.manuscriptcentral.com/photos> and edit your user information as appropriate.

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Thank you for submitting your manuscript to *Photosynthetica*.

Sincerely,  
Photosynthetica Editorial Office

Dear Editor-in-Chief of Photosynthetica and Reviewers

Thank you very much for reading our manuscript and giving us many kind advices. We would like to show our responses to editor and reviewers here, and provide the revised manuscript.

The blue color and additional of comments and supplementary file in the revised manuscript were in order to respond the suggestions of 1<sup>st</sup>, 2<sup>nd</sup> reviewers and associate editor.

Our explanation for the reviewers and associate editor comments is as follows.

Best regards

Akihiro Nose, Prof. Dr.  
Corresponding Author

#### **Explanatios of the reviewers' comments:**

**Reviewer #1:** In this manuscript, the authors simultaneously measured rates of photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption of *Rhizophora mucronata* leaves using conventional Clark-type O<sub>2</sub> electrode and CO<sub>2</sub> sensor under aqueous condition. They examined effects of sample status (slice or chip), pH and NaHCO<sub>3</sub> concentration on both rates. They showed that the photosynthetic quotient (PQ, O<sub>2</sub> evolution to CO<sub>2</sub> consumption) values were always higher than one, and suggested that photorespiration did not occur under this condition. Some data were interested. But firstly they should show the raw data of changes in O<sub>2</sub> and CO<sub>2</sub> concentrations in the cuvette using the O<sub>2</sub> electrode of Hansatech and the CO<sub>2</sub> sensor of Presens as the supplementary data.

>> Can both sensors similarly respond to the gas concentration in the cuvette? Was there any time lag between the changes of concentrations?

**Explanation:** Thank you for the valuable comments and we are also grateful for the time and energy you expended on our behalf. We agree with your suggestion. As our responses, we show the raw data of changes in O<sub>2</sub> and CO<sub>2</sub> concentrations in the cuvette using the O<sub>2</sub> electrode of Hansatech and the CO<sub>2</sub> sensor of Presens especially related with time lag as the supplementary data (see **Page 15, Supplementary 1**).

- Yes, there was the lag period between the changes of O<sub>2</sub> and CO<sub>2</sub> concentrations. We found the maximum lag period for CO<sub>2</sub> consumption was less than 2 min, but was about 3 min for O<sub>2</sub> evolution (see **Page 15, Supplementary 1**). Generally, lag period of O<sub>2</sub> evolution after light activation was slightly longer than CO<sub>2</sub> consumption but not in significance level. Furthermore, The O<sub>2</sub> evolution and CO<sub>2</sub> consumption rates were calculated from the initial slopes of the curves during linear photosynthetic activity after lag period finished (after around 4 min).

>> Also, why did they measure the photosynthetic rate at 25°C? Is it a bit low for this species?

**Explanation:** Moore et al (1973) reported that  $P_{\max}$  of mangrove *Rhizophora* and *Laguncularia* was obtained at leaf temperature near or below 25 °C. However, some latter reports indicate that the relationship between the maximum photosynthetic rate and leaf temperature indicated a wide peak between 29 and 34 °C (Okimoto et al. 2007). Therefore, next step we consider and also want to proceed more detail response of mangrove photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption rate simultaneously at 30°C.

**Reviewer #2:** An improved method for simultaneous determination of photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption of *Rhizophora mucronata* leaves Tengku et al. The paper compares measurement of photosynthesis by traditional gas exchange with liquid phase O<sub>2</sub> electrodes in combination of CO<sub>2</sub> optodes. The innovative method seems to function properly and provide comparative measurements to those from gas exchange at high light but not at low light. The paper is within the scope of *Photosynthetica*, and it is an interesting one. The best recommendation will be to test the method further before to publish the results and may be the paper will become in a Regular one and not just a Brief Communication. In case the Editor y other reviewers agree that paper can be published, which is acceptable as well, I have some suggestions to improve the text.

>> Introduction: Two and half pages are too long for a brief communication. It should be cut substantially.

**Explanation:** Thank you very much for your kind suggestions about our paper writing. We agree that some statements which are not important substantially should be cut. As our responses, we make some revisions:

- We omit statement “On these occasions, independent biochemical studies and photosynthesis analyses have been simultaneously conducted to estimate the effects of stress on leaf photosynthesis.” (see **Page 1, Line 11**)

- We delete “Furthermore, the possibility of directly determining the O<sub>2</sub> evolution and CO<sub>2</sub> consumption rate of leaf samples simultaneously and to monitor how rates change in response to stimuli will add to our understanding of single leaf to complex ecosystem mechanisms (Strovas *et al.* 2010).” (see **Page 1, Line 11**) and also omit Strovas *et al.* (2010) from references list (see **Page 13, Line 277**)

- We delete statements “The PQ value provides fundamental information on metabolic pathways (Taddei *et al.* 2008), balanced growth (Davies *et al.* 2003) and useful to clarify the primary productivity of an ecosystem (Lee and Bong 2006).” (See **Page 1, Line 23**) and also delete it from references list (see **Page 10, Line 221; Page 11 Line 247**).

- We omit reference “They belong to the C<sub>3</sub> photosynthetic group of plants, but may also be classified as “seaweeds”, since they can grow in submerged and highly saline conditions (Kawamitsu *et al.* 2003).” (See **Page 2, Line 31**) and also delete it from references list (see **Page 11, Line 243**).

>> Materials and Methods: The text should be arranged tidily to easy the understanding of this section.

**Explanation:** We add some information in the Material and Methods (blue color) to make it understanding easily.

>> Results and discussion: It requires to be shortened as well. I suggest that the final paragraph is arranged in a way that final take home part of the paper is the positive side. It is

good to write that the methods requires further improvements but I am not sure that should be the last message in the paper.

**Explanation:** We agree with your suggestion. As our responses, we close the final paragraph with the positive side (see **Page 9, Line 203-205**) and does not make the statement “the methods requires further improvements” as the last message in this manuscript (see **Page 8, Line 188-190**).

>> Figure, Table and References are Ok.

**Explanation:** Thank you very much for your kind words about our manuscript.

**Associate Editor:** Both reviewers are positive. Please revise the manuscript incorporating their suggestions. The editor also has some comments.

>> Please clarify that the authors used a liquid phase electrode. What was a closed chamber using an aqueous phase of the leaf disc oxygen electrode? In Line 174, they also stated that they used the leaf-disc oxygen electrode.

**Explanation:** Thank you for the valuable corrections and suggestions. We already changed leaf-disc oxygen electrode with a Clark oxygen electrode type polarographic sensor (see **Page 3, Line 56-57; Page 7, line 167-168; and Abstract, line 29-30**).

>> Please clearly state the oxygen concentration? The buffer was equilibrated with the air (21% O<sub>2</sub>)?

**Explanation:** The sensor and buffer was equilibrated with saturation air 21% and also zero oxygen line by using nitrogen bubble. We add this information in Material and method (see **Page 3, Line 68-71**).

>> The respiration rates in the dark (Figure 2) were grater in the liquid phase measurements than that obtained in the air measurement. Also the respiratory quotient (CO<sub>2</sub>/O<sub>2</sub>) was lower than 1.0. Was this reproducible? Perhaps, the greater respiratory rates were due to wounding?

**Explanation:** Although cutting leaves into small pieces result produce high O<sub>2</sub> evolution, we also agree that the wounding also due to O<sub>2</sub> consumption that will increase respiratory rate.

>> Because the photosynthesis quotients obtained were always greater than 1.0, the authors claimed that photorespiration was suppressed. However, the dependence of the photosynthetic rate on the bicarbonate concentrations indicates that the photosynthesis was not saturated by 5 mM or 10 mM bicarbonate ion. Thus, at these concentrations photorespiration would occur. However, the PQ values were also above 1.0? This editor noticed error bars were greater at low bicarbonate. Could the authors explain the discrepancy?

**Explanation:** Yes, beside low performance in low light condition, the other obstacle of our method is the demand for high carbon dioxide source from bicarbonate (see **Page 7, Line 144-145**). The PQ values above 1.0 were representative of mean values. However, some measurement also gain PQ values bit lower than 1.0.

>> If photorespiration is completely suppressed like the measurement in 5% CO<sub>2</sub> level, the rate of photosynthesis will be much greater than the rate

measured in 370  $\mu\text{mol mol}^{-1}$   $\text{CO}_2$  in air. It is necessary to explain the similar maximum values in Figure 2.

**Explanation:** yes, in comparison with gas exchange method, the maximum photosynthetic rate in photosynthetic  $\text{O}_2$  evolution and  $\text{CO}_2$  consumption was achieved under very high carbon dioxide condition. We add this information in result and discussion (see **Page 9, Line 199-201**).

>> There are some technical problems as well. For all modifications, follow „Instructions for authors“ and “Detailed instructions for preparation of papers“ carefully.

**Explanation:** We try to improve our manuscript writing based on “Instructions for authors“ and “Detailed instructions for preparation of papers“.

1 Leaf O<sub>2</sub> evolution and CO<sub>2</sub> consumption are fundamental mechanisms that support  
2 oxygen and carbon ecosystems from the individual plant to the global scale. Based on  
3 the photosynthesis chemical formula, which justifies that the ratio of O<sub>2</sub> evolution to  
4 CO<sub>2</sub> fixation is 1:1 (Espie 1986), the traditional estimation of photosynthetic gas  
5 exchange has been evaluated either by O<sub>2</sub> evolution or CO<sub>2</sub> consumption. However, in  
6 an intact leaf, some physiological functions that synthesise and consume O<sub>2</sub> and CO<sub>2</sub>  
7 may vary, particularly under stress conditions (Wu *et al.* 2014), photorespiration  
8 (Rosenberg *et al.* 1995) and other oxygenative functions (Taddei *et al.* 2008). This  
9 means that the ratio of O<sub>2</sub>:CO<sub>2</sub> during photosynthesis in intact leaves is not always 1:1.

10 The simultaneous estimation of O<sub>2</sub> and CO<sub>2</sub> has been done using isotope-Gas  
11 Chromatography-Mass Spectrometry (GC-MS) with <sup>13</sup>CO<sub>2</sub> and <sup>18</sup>O<sub>2</sub> (Isobe *et al.* 2011).  
12 However, the method is unpopular because the equipment is very expensive (Sipior *et*  
13 *al.* 1996). In this study, we tried to improve the potential for a convenient evaluation of  
14 O<sub>2</sub> evolution and CO<sub>2</sub> consumption in photosynthesis by using an O<sub>2</sub> electrode and CO<sub>2</sub>  
15 optodes simultaneously. The main advantages of optodes are that they can be used in  
16 non-invasive systems, oxygen and carbon dioxide are not consumed by the optodes,  
17 measurements are possible over a wide temperature range, and there is no mechanical  
18 stress (Warkentin *et al.* 2007). If this simultaneous method is convenient, it becomes a  
19 useful mechanism to more easily study the physiological effects of photosynthesis.

20 A simultaneous measurement of O<sub>2</sub> evolution and CO<sub>2</sub> consumption during  
21 photosynthesis is also essential in order to calculate the photosynthetic quotient (PQ),  
22 which is described as the molar ratio of the rate of O<sub>2</sub> production to the rate of CO<sub>2</sub>  
23 utilization (Williams and Robertson 1991). Some ecosystem productivity studies  
24 have been made with the assumption that PQ = 1 (Suzumura *et al.* 2002, Nielsen

**Commented [Zia1]:** deleted Furthermore, the possibility of directly determining the O<sub>2</sub> evolution and CO<sub>2</sub> consumption rate of leaf samples simultaneously and to monitor how rates change in response to stimuli will add to our understanding of single leaf to complex ecosystem mechanisms (Strovas *et al.* 2010).

**Commented [Zia2]:** deleted  
On these occasions, independent biochemical studies and photosynthesis analyses have been simultaneously conducted to estimate the effects of stress on leaf photosynthesis.

**Commented [Zia3]:** deleted  
The PQ value provides fundamental information on metabolic pathways (Taddei *et al.* 2008), balanced growth (Davies *et al.* 2003) and useful to clarify the primary productivity of an ecosystem (Lee and Bong 2006).

25 and Nielsen 2006) that could affect data interpretation of tropical productivity  
26 (Taddei *et al.* 2008).

27 Mangroves represent an important coastal ecosystem in tropical areas. During the  
28 seedling stage, the red mangrove (*Rhizophora mucronata* L.) lives periodically in  
29 submerged conditions like seaweed or macroalgae. Our previous work (Ulqodry *et al.*  
30 2014) explored the photosynthetic performance of *R. mucronata* leaves using the gas  
31 exchange method. This method had a high precision and was rapid (Moore *et al.* 1973,  
32 Sobrado 2005, Okimoto *et al.* 2007) but was limited under aqueous conditions as the  
33 Infra-Red Gas Analyser is sensitive to water immersion (Gevaert *et al.* 2011). The  
34 advent of a new type of optical electrodes, the so-called opt(r)odes, facilitated the  
35 estimation of the *R. mucronata* photosynthetic rate under aqueous conditions. Previous  
36 studies have applied optodes for oxygen and carbon independently in the water column,  
37 sediments and plant tissues (Gansert *et al.* 2001, Glud *et al.* 2005, Berggren *et al.* 2012).

38 In this study, we examined the photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption  
39 rates of *R. mucronata* leaves under aqueous condition. In a simultaneous experiment we  
40 used a liquid-phase O<sub>2</sub> electrode and CO<sub>2</sub> optodes to demonstrate their interdependence  
41 and differences and compared the results with those of the gas exchange method. The  
42 determination of PQ values and light-saturated photosynthetic rate ( $P_{\max}$ ) of *R.*  
43 *mucronata* under aqueous conditions was investigated.

44 Propagules of *R. mucronata* were obtained from a mangrove area on Galang  
45 Island, Batam District, Indonesia (0° 45' N, 104° 15' E). Propagules were initially grown  
46 in a heated greenhouse at the Laboratory of Tropical Crop Improvement, Saga  
47 University, Japan (33° 14' N, 130° 17' E). The fully expanded leaves from 3–4  
48 mangrove seedlings were used as materials.

**Commented [Zia4]:** deleted: They belong to the C<sub>3</sub> photosynthetic group of plants, but may also be classified as "seaweeds", since they can grow in submerged and highly saline conditions (Kawamitsu *et al.* 2003).

49 Leaves were collected early each morning, vacuum-infiltrated with the buffer and  
50 stored in the dark until required. One essential consequence of this treatment was the  
51 inactivation of rubisco, so that the photosynthetic rates were approximately 10% of  
52 those generally observed from leaves taken directly from a plant (Brown 1998). The leaf  
53 sample was sliced into squares of approximately 1 mm<sup>2</sup>. The leaves were sliced under a  
54 50 mM HEPES buffer containing 0.5 mM CaSO<sub>4</sub> and transferred into the electrode  
55 chamber that contained the same buffer.

56 Photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption were measured simultaneously  
57 in a closed chamber using an aqueous phase of a Clark oxygen electrode type  
58 polarographic sensor (Hansatech, Norfolk, UK) with a 'pCO<sub>2</sub> mini' optodes sensor  
59 (PreSens GmbH, Regensburg, Germany) that was inserted into the chamber. The  
60 optodes system guarantees a high temporal resolution and a measurement without drift,  
61 oxygen consumption, or gas exchange between the incubation chamber and the  
62 environment (Warkentin *et al.* 2007). The chamber was equipped with a water jacket to  
63 maintain the temperature at 25°C. Periodic checking ensured that the highest  
64 illumination intensity did not result in a rapid increase in temperature. Light was  
65 provided by a slide projector lamp and the lens system focussed the light into the  
66 electrode compartment. The photosynthetically active radiation (PAR) in the chamber  
67 was measured with a quantum sensor (*model QRT1, Hansatech, Norfolk, UK*). It is  
68 important that the slices do not obstruct the rotation of the magnetic flea and also the  
69 sensor of pCO<sub>2</sub> mini. To achieve maximum accuracy, a two point calibration of sensor  
70 and buffer was equilibrated with saturation air 21% and also zero oxygen line by using  
71 nitrogen bubble. This process also removed any dissolved CO<sub>2</sub> from the medium, such  
72 that the added NaHCO<sub>3</sub> was the only carbon source available.



73           Photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption of *R. mucronata* leaves under  
74 aqueous conditions were measured at various pH levels, NaHCO<sub>3</sub> concentrations and  
75 PAR levels at a temperature of 25°C. The relationship between the pH of the buffer and  
76 apparent photosynthetic rate was measured at pH 6.0, 6.5, 7.0, 7.5, 8.0, and 9.0 with  
77 NaHCO<sub>3</sub> 20 mM as carbon dioxide source under saturated PAR 1,000 μmol m<sup>-2</sup> s<sup>-1</sup>.  
78 The effect of different NaHCO<sub>3</sub> concentrations (0, 5, 10, 20 and 40 mM) was measured  
79 at pH 7.5 and a saturated PAR of 1,000 μmol m<sup>-2</sup> s<sup>-1</sup>. In relation to light intensity, PAR  
80 values in the chamber were maintained in decreasing levels from 1,000 to 50 μmol m<sup>-2</sup>  
81 s<sup>-1</sup> by placing various distance between projector lamp and the chamber. For a dark  
82 respiration measurement, the electrode chamber was wrapped in two layers of  
83 aluminium foil.

84           The O<sub>2</sub> electrode signal was recorded using Oxygraph Plus System software  
85 (*Hansatech*, Norfolk, UK) as a real-time chart recorder simulation. Simultaneously, the  
86 CO<sub>2</sub> consumption was measured in the same chamber every 5 s using pCO<sub>2</sub> View  
87 v1.0.2 software (*PreSens GmbH*, Regensburg, Germany). There was a lag period less  
88 than 2 min for CO<sub>2</sub> consumption, and about 3 min for O<sub>2</sub> evolution after light activation  
89 (Supplementary 1). Generally, lag period of O<sub>2</sub> evolution was slightly longer than CO<sub>2</sub>  
90 consumption but not in significance level. Furthermore, The O<sub>2</sub> evolution and CO<sub>2</sub>  
91 consumption rates were calculated from the initial slopes of the curves during linear  
92 photosynthetic activity after lag period finished.

93           As a comparison, the photosynthetic rate based on gas exchange in the air was  
94 also performed on leaf pairings similar to those used to measure O<sub>2</sub> evolution and CO<sub>2</sub>  
95 consumption under aqueous conditions. Measurements of leaf gas exchange were  
96 conducted using a portable open-flow gas exchange system (*LI-6400*, *Li-COR*, Lincoln,

97 NE, USA). The effect of light intensity on the photosynthetic rate was measured from  
98 PAR 1,000 to 0  $\mu\text{mol m}^{-2} \text{s}^{-1}$  (1,000; 500, 250, 100, 50, 0  $\mu\text{mol m}^{-2} \text{s}^{-1}$ ) under leaf  
99 temperature, VpdL and  $\text{CO}_2$  input were  $25^\circ\text{C}$ ,  $1.7 \pm 0.3 \text{ kPa}$ , and  $370 \mu\text{mol mol}^{-1}$ ,  
100 respectively. The light responses of the photosynthetic rate was determined using  
101 the rectangular hyperbola model (Okimoto *et al.* 2008) to specify the  $P_{\text{max}}$  of *R.*  
102 *mucronata* leaves (Ulqodry *et al.* 2014) in air and under aqueous conditions.

103 Analysis of variance (ANOVA) was performed using statistiXL Version 1.x.  
104 Significant differences between treatments were further evaluated using the Tukey HSD  
105 test ( $P < 0.05$ ).

106 We began the experiment by comparing the most suitable leaf shape that resulted  
107 in highest  $\text{O}_2$  evolution and  $\text{CO}_2$  consumption, between small slice pieces ( $1 \text{ mm}^2$ ) and a  
108 larger, chip shape ( $1 \text{ cm}^2$ ). Previous results indicated that cutting leaves into small  
109 pieces can be negligible during  $\text{O}_2$  evolution measurement under aqueous condition  
110 (Kawamitsu and Boyer 1999). Our results showed that a small *R. mucronata* leaf  
111 sample had significantly had higher  $\text{O}_2$  evolution and  $\text{CO}_2$  consumption rates compared  
112 with the larger, chip shape (Fig. 1-*I*). This suggests that slicing the leaf tissues facilitates  
113 the increasing of gas exchange across the boundary layer at the tissue surface (Brown  
114 1998). This eliminates the effect of stomatal resistance for  $\text{CO}_2$  diffusion, and free  $\text{CO}_2$   
115 molecules or  $\text{HCO}_3^-$  ions may penetrate more easily into the tissue of the leaf slice,  
116 resulting in a higher photosynthetic rate (Ishii *et al.* 1977).

117 The most important factors for measuring net photosynthetic rate ( $P_N$ ) in aqueous  
118 conditions are the pH and carbonate system of the reaction mixture. Dissolved carbon  
119 dioxide in water occurs in three inorganic forms, free aqueous carbon dioxide (free  
120  $\text{CO}_2$ ), bicarbonate ( $\text{HCO}_3^-$ ) and carbonate ions ( $\text{CO}_3^{2-}$ ). If the equilibrium is affected by

121 a change in pH, this could potentially influence  $P_N$  (Riebesell *et al.* 2007). The  $P_N$  in  
122 response to pH exhibited a similar pattern for both  $O_2$  evolution and  $CO_2$  consumption,  
123 with higher with associated with intermediate pH values of 7.0–7.5 compared with low  
124 and high pH (Fig. 1-II). Under a high pH condition of 8.0–9.0, free molecular  $CO_2$   
125 decreased and bicarbonate increased (Chen and Durbin 1994). This meant that the free  
126  $CO_2$  in the reaction mixture became limiting, reducing  $P_N$ . This result also demonstrated  
127 that the main carbon utilised as the substrate for *R. mucronata* leaf photosynthesis was  
128 free  $CO_2$  molecules rather than bicarbonate. Almost all terrestrial plants use only free  
129  $CO_2$  for photosynthesis, however, many seaweeds or macroalgae use both free  $CO_2$  and  
130 external bicarbonate in water as a source of carbon for photosynthesis (Kawamitsu and  
131 Boyer 1999, Pierini and Thomaz 2004). However, the photosynthetic rate that was  
132 detected, even at a pH of 9, indicated that *R. mucronata* leaves used bicarbonate as an  
133 additional source of carbon under low free  $CO_2$  conditions. The requirement of  
134 Photosystem II (PSII) for bicarbonate (carbonate) has been observed for intact leaves,  
135 isolated thylakoids and PSII-enriched membrane fragments from oxygenic  
136 photosynthesisers (Shevela *et al.* 2012). Bicarbonate is required for the regulation of  
137 photosynthetic electron transport on the acceptor side of PSII (Wydrzynski and  
138 Govindjee 1975), and is probably also involved in the mechanism of  $O_2$  evolution  
139 on the oxidising side of PSII (Stemler 2002).

140 This study was unable to demonstrate that high free  $CO_2$  under low pH  
141 condition ( $< 7.0$ ) resulted in a high  $P_N$ . It described that leaf  $O_2$  evolution and  $CO_2$   
142 consumption were strongly related to leaf intracellular conditions. Berge *et al.* (2010)  
143 pointed out that as the pH dropped the  $H^+$  concentration increased which may affect  
144 intracellular pH, which may affect intracellular pH, membrane potential, energy

145 partitioning, and enzyme activity. For this reason, aqueous acidification may reduce  $P_N$   
146 through direct pH effects.

147 Variation in  $P_N$  responses to  $\text{NaHCO}_3$  concentrations also showed almost  
148 similar trends for both  $\text{O}_2$  evolution and  $\text{CO}_2$  consumption. The  $P_N$  increased with  
149 higher  $\text{NaHCO}_3$  concentrations until reaching the saturation point at 20 mM (Fig. 1-  
150 *III*). The high bicarbonat saturation point indicated that our method need very high  
151 carbondioxide source. Particularly in submerged plants,  $P_N$  may be limited by a low  
152 availability of dissolved inorganic carbon (Maberly and Spence 1983, Adamec  
153 1997). The interesting finding was that although there was no significant difference  
154 between  $\text{O}_2$  evolution and  $\text{CO}_2$  consumption,  $\text{O}_2$  evolution values were always higher  
155 than  $\text{CO}_2$  consumption values under the different pH and  $\text{NaHCO}_3$  concentrations. This  
156 result is important if we want to explore the PQ of *R. mucronata* leaves under aqueous  
157 conditions. To be useful, PQ should be determined using the net rate of  $\text{O}_2$  involved per  
158  $\text{CO}_2$  fixed simultaneously and can be described as,

$$159 \quad PQ = \frac{O_2 \text{ evolution}}{CO_2 \text{ consumption}}$$

160 Stoichiometrically, a PQ value equal to unity ( $PQ = 1.00$ ) assumes hexose  
161 production with ammonium as the N source (Rosenberg *et al.* 1995). If this simple  
162 photosynthesis physiology was replaced by an ecological summation of protoplasm  
163 production, including carbohydrates, protein, lipids, and nucleic acids, then the  
164 theoretical PQ would be higher (Williams and Robertson 1991). Theoretical PQ values  
165 typically range from 1.0–1.3 (Rosenberg *et al.* 1995). The PQ values of *R. mucronata*  
166 leaves in different pH and  $\text{NaHCO}_3$  concentrations ranged from 1.04–1.28 with no  
167 significant difference among them (Fig. 1-*IV*). Purely based on stoichiometric and  
168 theoretical considerations of PQ values, results similar or higher than unity would be

169 expected. A PQ of 1.0 infers that the sole product of photosynthesis is carbohydrate  
170 and a PQ > 1.0 indicates that more reduced compounds are produced, such as fats  
171 and proteins (Chisholm 1998). Our results also suggested that the simultaneous  
172 measurement of O<sub>2</sub> evolution and CO<sub>2</sub> consumption by using a Clark oxygen electrode  
173 type polarographic sensor and 'pCO<sub>2</sub> mini' optodes sensor provided a simple, stable  
174 and precise measurement of net PQ under aqueous conditions.

175 The net PQ values in all measurements was never less than unity and confirmed  
176 that photorespiration did not occur under aqueous conditions. A possible explanation for  
177 a PQ less than unity would be photorespiration (glycolate production) as a result of  
178 oxygenase activity of ribulose-1,5-bisphosphate (RuBP) carboxylase at high ambient  
179 oxygen concentrations (Rosenberg *et al.* 1995). Photorespiration that decreases PQ  
180 occurs when rubisco, which principally functions as carboxylase, is substituted by the  
181 oxygenase function (Taddei *et al.* 2008). In terrestrial C<sub>3</sub> plants, photorespiratory  
182 consumption of O<sub>2</sub> can account for 25% of rubisco activity (Falkowski and Raven 1997).  
183 Conversely, photorespiration is assumed to be of minor importance to aquatic plants  
184 compared with terrestrial C<sub>3</sub> plants (Laws *et al.* 2000), because submerged  
185 environmental conditions, such as fairly constant oxygen and total inorganic carbon  
186 concentrations, does not favour photorespiration (Rosenberg *et al.* 1995).

187 In order to characterise the functioning of photosynthetic apparatus of *R.*  
188 *mucronata* in air and aqueous conditions, the light curves of P<sub>N</sub> for similarly paired  
189 leaves were estimated. In Fig. 2, at low light levels (PAR < 500 μmol photons m<sup>-2</sup> s<sup>-1</sup>),  
190 the photosynthetic rate of O<sub>2</sub> evolution and CO<sub>2</sub> consumption under aqueous conditions  
191 was lower than the photosynthetic CO<sub>2</sub> exchange in air. This result is likely to be related  
192 to the reduction of low light utilisation while the leaf slices were rotated under aqueous

193 conditions. Another possible explanation for this was that our method worked well  
194 under light saturation compared with light limitation. Therefore, we need to improve the  
195 simultaneous measurements of photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption under  
196 aqueous conditions in low light conditions.

197 The light saturation points for all  $P_N$  measurements (CO<sub>2</sub> exchange in air, O<sub>2</sub>  
198 evolution under aqueous condition and CO<sub>2</sub> consumption under aqueous condition)  
199 were similar at PAR levels around 500–1,000  $\mu\text{mol photons m}^{-2} \text{ s}^{-1}$ . The  $P_{\text{max}}$  which  
200 demonstrates the potential photosynthetic capacity of *R. mucronata* leaves (Ulqodry *et*  
201 *al.* 2014), was also determined. All experiments produced comparable results with  
202 similar  $P_{\text{max}}$  values of 13.37, 13.11 and 12.31  $\mu\text{mol m}^{-2} \text{ s}^{-1}$  for CO<sub>2</sub> exchange in air, O<sub>2</sub>  
203 evolution under aqueous conditions and CO<sub>2</sub> consumption under aqueous conditions,  
204 respectively. In comparison with gas exchange, the maximum photosynthetic rate in  
205 photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption under aqueous condition was achieved under  
206 very high carbon dioxide condition. The  $P_{\text{max}}$  value and daily period of irradiance when  
207 plants were in the water and air would be useful as an indicator of primary production  
208 (Zimmerman *et al.* 1994). The similar  $P_{\text{max}}$  values suggested that all treatments resulted  
209 in a high capacity to adjust the photosynthetic apparatus under light saturation  
210 conditions.

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## BRIEF COMMUNICATION

## An improved method for the simultaneous determination of photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption in *Rhizophora mucronata* leaves

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### Abstract

The photosynthetic gas exchange has been assessed traditionally either as O<sub>2</sub> evolution or CO<sub>2</sub> consumption. In this study, we used a liquid-phase O<sub>2</sub> electrode combined with CO<sub>2</sub> optodes to examine simultaneously photosynthesis in intact leaves of mangrove *Rhizophora mucronata*. We verified suitable conditions for leaf photosynthetic rates by assessing pH levels and NaHCO<sub>3</sub> concentrations and compared these to the gas exchange method at various PAR levels. The photosynthetic rate in response to pH exhibited a similar pattern both for O<sub>2</sub> evolution and CO<sub>2</sub> consumption, and higher rates were associated with intermediate pH compared with low and high pH values. The net photosynthetic quotient (PQ) of *R. mucronata* leaves ranged from 1.04–1.28. The PQ values, which were never lesser than 1, suggested that photorespiration did not occur in *R. mucronata* leaves under aqueous conditions. The similar maximum photosynthetic rates suggested that all measurements had a high capacity to adjust to not clear the photosynthetic apparatus under a light saturation condition. The simultaneous measurements of O<sub>2</sub> evolution and CO<sub>2</sub> consumption using the Clark oxygen electrode polarographic sensor with the CO<sub>2</sub> optode sensor provided a simple, stable, and precise measurement of PQ under aqueous and saturated light conditions.

*Additional key words:* carbon dioxide consumption; oxygen evolution; photosynthetic performance.

Leaf O<sub>2</sub> evolution and CO<sub>2</sub> consumption are fundamental mechanisms that support oxygen and carbon ecosystems from the individual plant to the global scale. Based on the photosynthesis chemical formula, which justifies that the ratio of O<sub>2</sub> evolution to CO<sub>2</sub> fixation is 1:1 (Espie 1986), the traditional estimation of photosynthetic gas exchange has been evaluated either by O<sub>2</sub> evolution or CO<sub>2</sub> consumption. However, in an intact leaf, some physiological functions that synthesise and consume O<sub>2</sub> and CO<sub>2</sub> may vary, particularly under stress conditions (Wu *et al.* 2014), photorespiration (Rosenberg *et al.* 1995), and other oxygenative functions (Taddei *et al.* 2008). This means that the ratio of O<sub>2</sub>:CO<sub>2</sub> during photosynthesis in intact leaves is not always 1:1.

The simultaneous estimation of O<sub>2</sub> and CO<sub>2</sub> has been done using isotope-gas chromatography-mass spectrometry (GC-MS) with <sup>13</sup>CO<sub>2</sub> and <sup>18</sup>O<sub>2</sub> (Isobe *et al.* 2011).

However, the method is unpopular because the equipment is very expensive (Sipior *et al.* 1996). In this study, we tried to improve the potential for a convenient evaluation of O<sub>2</sub> evolution and CO<sub>2</sub> consumption in photosynthesis by using the O<sub>2</sub> electrode and CO<sub>2</sub> optodes simultaneously. The main advantages of optodes are that they can be used in non-invasive systems, oxygen and carbon dioxide are not consumed by the optodes, measurements are possible over a wide temperature range, and there is no mechanical stress (Warkentin *et al.* 2007). If this simultaneous method is convenient, it becomes a useful mechanism to study more easily physiological effects of photosynthesis.

A simultaneous measurement of O<sub>2</sub> evolution and CO<sub>2</sub> consumption during photosynthesis is also essential in order to calculate the photosynthetic quotient (PQ), which is described as the molar ratio of the rate of O<sub>2</sub>

Received 8 April 2015, accepted 22 June 2015.

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*Abbreviations:* P<sub>max</sub> – light-saturated photosynthetic rate; P<sub>N</sub> – net photosynthetic rate; PQ – photosynthetic quotient; RuBP – ribulose-1,5-bisphosphate.

*Acknowledgements:* We acknowledge The United Graduate School of Agricultural Sciences, Kagoshima University, Japan for continuous funding through The Rendai Research Subsidies Program. We also thank Yutaro Oba for his technical support.

production to the rate of CO<sub>2</sub> utilization (Williams and Robertson 1991). Some ecosystem productivity studies have been made with the assumption that PQ = 1 (Suzumura *et al.* 2002, Nielsen and Nielsen 2006); it could affect data interpretation of tropical productivity (Taddei *et al.* 2008).

Mangroves represent an important coastal ecosystem in tropical areas. During the seedling stage, red mangrove (*Rhizophora mucronata* L.) lives periodically in submerged conditions similar to seaweed or macroalgae. Our previous work (Ulqodry *et al.* 2014) explored the photosynthetic performance of *R. mucronata* leaves using the gas-exchange method. This method showed a high precision and was rapid (Moore *et al.* 1973, Sobrado 2005, Okimoto *et al.* 2007), but it was limited under aqueous conditions as the infra-red gas analyser was sensitive to water immersion (Gevaert *et al.* 2011). The advent of a new type of optical electrodes, the so-called opt(r)odes, facilitated the estimation of the *R. mucronata* photosynthetic rate under aqueous conditions. Previous studies have applied optodes for oxygen and carbon independently in the water column, sediments, and plant tissues (Gansert *et al.* 2001, Glud *et al.* 2005, Berggren *et al.* 2012).

In this study, we examined the photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption rates of *R. mucronata* leaves under aqueous condition. In a simultaneous experiment, we used a liquid-phase O<sub>2</sub> electrode and CO<sub>2</sub> optodes to demonstrate their interdependence and differences and compared the results with those of the gas-exchange method. The determination of PQ values and light-saturated photosynthetic rate ( $P_{\max}$ ) of *R. mucronata* under aqueous conditions was investigated.

Propagules of *R. mucronata* were obtained from a mangrove area on the Galang Island, Batam District, Indonesia (0°45'N, 104°15'E). Propagules were initially grown in a heated greenhouse at the Laboratory of Tropical Crop Improvement, Saga University, Japan (33°14'N, 130°17'E). The fully expanded leaves from 3–4 mangrove seedlings were used as materials.

The leaves were collected in early morning, vacuum-infiltrated with the buffer, and stored in the dark until required. One essential consequence of this treatment was the inactivation of Rubisco, so that the photosynthetic rates were approximately 10% of those generally observed from leaves taken directly from a plant (Brown 1998). The leaf sample was sliced into squares of approximately 1 mm<sup>2</sup>. The leaves were sliced under a 50 mM HEPES buffer containing 0.5 mM CaSO<sub>4</sub> and transferred into the electrode chamber that contained the same buffer.

Photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption were measured simultaneously in a closed chamber using an aqueous phase of a Clark oxygen electrode type polarographic sensor (Hansatech, Norfolk, UK) with a 'pCO<sub>2</sub> mini' optodes sensor (PreSens GmbH, Regensburg, Germany) that was inserted into the

chamber. The optodes system guarantees a high temporal resolution and a measurement without drift, oxygen consumption, or gas exchange between the incubation chamber and the environment (Warkentin *et al.* 2007). The chamber was equipped with a water jacket to maintain temperature at 25°C. Periodic checking ensured that the highest illumination intensity did not result in a rapid increase in temperature. Light was provided by a slide projector lamp and the lens system focused the light into the electrode compartment. The photosynthetically active radiation in the chamber was measured with a quantum sensor (model QRT1, Hansatech, Norfolk, UK). It was important that the slices did not obstruct the rotation of the magnetic flea and also the sensor of pCO<sub>2</sub> mini. To achieve maximum accuracy, a two point calibration of the sensor and buffer was equilibrated with saturation air 21% and also zero oxygen line by using nitrogen bubble. This process also removed any dissolved CO<sub>2</sub> from the medium, such that the added NaHCO<sub>3</sub> was the only carbon source available.

Photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption of *R. mucronata* leaves under aqueous conditions were measured at various pH levels, NaHCO<sub>3</sub> concentrations, and PAR levels at temperature of 25°C. The relationship between the pH of the buffer and apparent photosynthetic rate was measured at pH 6.0, 6.5, 7.0, 7.5, 8.0, and 9.0 with 20 mM NaHCO<sub>3</sub> as carbon dioxide source under saturation PAR 1,000 μmol(photon) m<sup>-2</sup> s<sup>-1</sup>. The effect of different NaHCO<sub>3</sub> concentrations (0, 5, 10, 20, and 40 mM) was measured at pH 7.5 and a saturation PAR of 1,000 μmol(photon) m<sup>-2</sup> s<sup>-1</sup>. In relation to light intensity, PAR values in the chamber were maintained in decreasing levels from 1,000 to 50 μmol(photon) m<sup>-2</sup> s<sup>-1</sup> by placing various distance between the projector lamp and the chamber. For a dark respiration measurement, the electrode chamber was wrapped in two layers of aluminium foil.

The O<sub>2</sub> electrode signal was recorded using *Oxygraph Plus System* software (Hansatech, Norfolk, UK) as a real-time chart recorder simulation. Simultaneously, the CO<sub>2</sub> consumption was measured in the same chamber every 5 s using *pCO<sub>2</sub> View v1.0.2* software (PreSens GmbH, Regensburg, Germany). There was a lag period lesser than 2 min for CO<sub>2</sub> consumption, and about 3 min for O<sub>2</sub> evolution after light activation (Fig. 1S, *supplement available online*). Generally, the lag period of O<sub>2</sub> evolution was slightly longer than that of CO<sub>2</sub> consumption, but insignificantly. Furthermore, the O<sub>2</sub> evolution and CO<sub>2</sub> consumption rates were calculated from the initial slopes of the curves during a linear photosynthetic activity rate after the lag period finished.

As a comparison, the photosynthetic rate based on gas exchange in the air was also performed on leaf pairings similar to those used to measure O<sub>2</sub> evolution and CO<sub>2</sub> consumption under aqueous conditions. Measurements of leaf gas exchange were conducted using a portable open-flow gas exchange system (LI-6400, Li-COR, Lincoln,

NE, USA). The effect of light intensity on the photosynthetic rate was measured from PAR 1,000 to 0  $\mu\text{mol}(\text{photon})\text{ m}^{-2}\text{ s}^{-1}$  (1,000; 500, 250, 100, 50, 0  $\mu\text{mol}\text{ m}^{-2}\text{ s}^{-1}$ ) with leaf temperature,  $V_{\text{pdI}}$ ? and CO<sub>2</sub> input of 25°C,  $1.7 \pm 0.3$  kPa, and 370  $\mu\text{mol mol}^{-1}$ , respectively. The light responses of the photosynthetic rate was determined using the rectangular hyperbola model (Okimoto *et al.* 2008) to specify the  $P_{\text{max}}$  of *R. mucronata* leaves (Ulqodry *et al.* 2014) in air and under aqueous conditions.

Analysis of variance (ANOVA) was performed using *StatistiXL Version 1.x*. Significant differences between treatments were further evaluated using the *Tukey's HSD* test ( $P < 0.05$ ).

We began the experiment by comparing the most suitable leaf shape that would result in the highest O<sub>2</sub> evolution and CO<sub>2</sub> consumption, *i.e.*, small slice pieces (1 mm<sup>2</sup>) and a larger, chip shape ones (1 cm<sup>2</sup>). Previous results indicated that cutting leaves into small pieces can be negligible during O<sub>2</sub> evolution measurement under aqueous condition (Kawamitsu and Boyer 1999). Our results showed that the small *R. mucronata* leaf sample exhibited significantly higher O<sub>2</sub> evolution and CO<sub>2</sub> consumption rates to the larger, chip shape pieces (Fig. 1A). This suggests that slicing the leaf tissues

facilitated increasing gas exchange across the boundary layer at the tissue surface (Brown 1998). This eliminates the effect of stomatal resistance for CO<sub>2</sub> diffusion, and free CO<sub>2</sub> molecules or HCO<sub>3</sub><sup>-</sup> ions may penetrate more easily into the tissue of the leaf slice, resulting in a higher photosynthetic rate (Ishii *et al.* 1977).

The most important factors for measuring the net photosynthetic rate ( $P_{\text{N}}$ ) in aqueous conditions were pH and a carbonate system of the reaction mixture. Dissolved carbon dioxide in water occurs in three inorganic forms: free aqueous carbon dioxide (free CO<sub>2</sub>), bicarbonate (HCO<sub>3</sub><sup>-</sup>), and carbonate ions (CO<sub>3</sub><sup>2-</sup>). If the equilibrium is affected by a change in pH, this could potentially influence  $P_{\text{N}}$  (Riebesell *et al.* 2007). The  $P_{\text{N}}$  in response to pH exhibited a similar pattern for both O<sub>2</sub> evolution and CO<sub>2</sub> consumption, with higher with associated with intermediate not clear, rephrase pH values of 7.0–7.5 compared to low and high pH (Fig. 1B). Under a high pH condition of 8.0–9.0, free molecular CO<sub>2</sub> decreased and bicarbonate increased (Chen and Durbin 1994). This meant that free CO<sub>2</sub> in the reaction mixture became limiting and it reduced  $P_{\text{N}}$ . This result also demonstrated that the main carbon form utilised as the substrate for *R. mucronata* leaf photosynthesis was free CO<sub>2</sub> molecules rather than bicarbonate. Almost all terrestrial plants

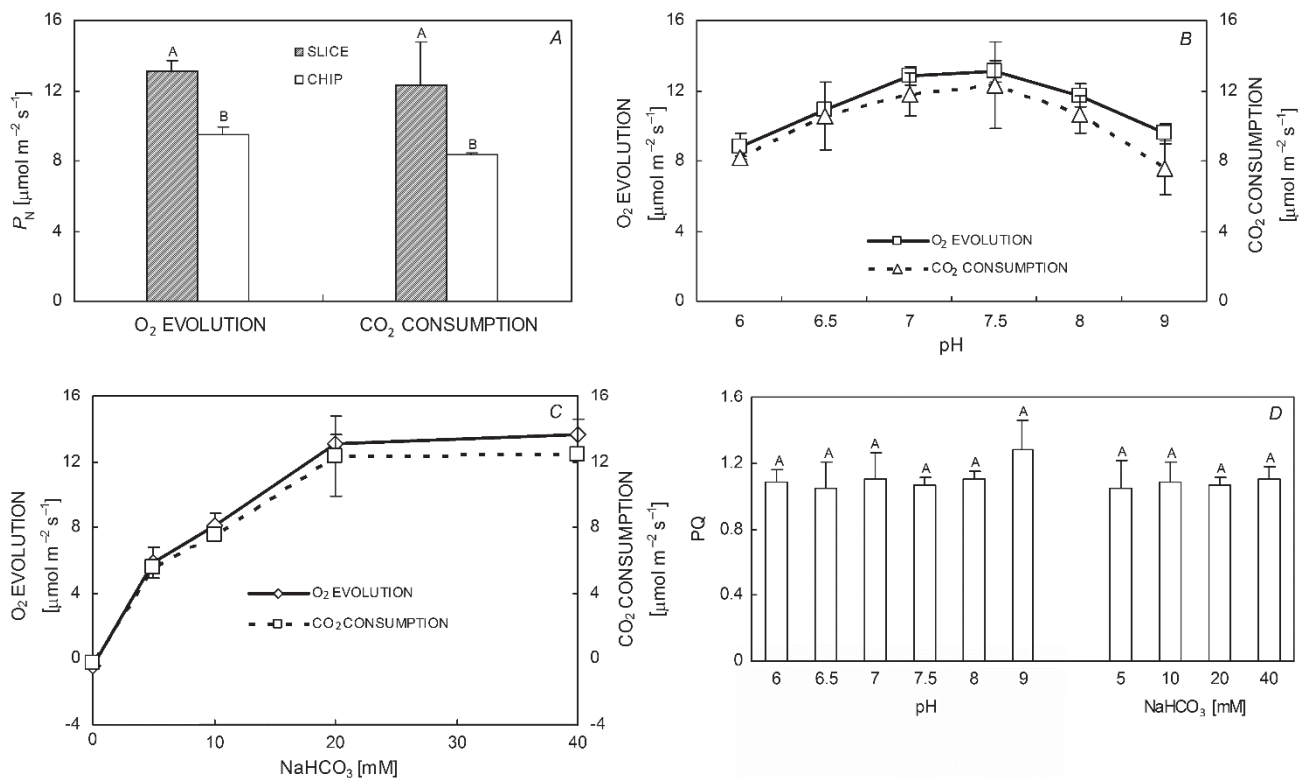


Fig. 1. Photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption of *R. mucronata* leaves under aqueous conditions measured in various: sample forms (A), pH levels (B), and NaHCO<sub>3</sub> concentrations (C); and the photosynthetic quotient (PQ) values as ratio of O<sub>2</sub> evolution and CO<sub>2</sub> consumption (D). The conditions of the measurements: temperature of 25°C and PAR of 1,000  $\mu\text{mol}(\text{photon})\text{ m}^{-2}\text{ s}^{-1}$ . Values are means  $\pm$  SD. Different letters over bars represent significant differences among all treatments ( $P < 0.05$ , *Tukey's HSD* test).

use only free CO<sub>2</sub> for photosynthesis, however, many seaweeds or macroalgae use both free CO<sub>2</sub> and external bicarbonate in water as the source of carbon for photosynthesis (Kawamitsu and Boyer 1999, Pierini and Thomaz 2004). However, the photosynthetic rate, which was detected even at pH 9, indicated that *R. mucronata* leaves used bicarbonate as an additional source of carbon under low free CO<sub>2</sub> conditions. The requirement of PSII for bicarbonate (carbonate) has been observed for intact leaves, isolated thylakoids, and PSII-enriched membrane fragments from oxygenic photosynthesisers (Shevela *et al.* 2012). Bicarbonate is required for the regulation of photosynthetic electron transport on the acceptor side of PSII (Wydrzynski and Govindjee 1975), and is probably also involved in the mechanism of O<sub>2</sub> evolution on the oxidising side of PSII (Stemler 2002).

This study was unable to demonstrate that high free CO<sub>2</sub> under low pH condition (<7.0) resulted in high  $P_N$ . It seemed that leaf O<sub>2</sub> evolution and CO<sub>2</sub> consumption were strongly related to leaf intracellular conditions. Berge *et al.* (2010) pointed out that as pH dropped the H<sup>+</sup> concentration increased which may affect intracellular pH, membrane potential, energy partitioning, and enzyme activity. For this reason, aqueous acidification may reduce  $P_N$  through direct pH effects.

Variation in  $P_N$  responses to NaHCO<sub>3</sub> concentrations also showed almost similar trends for both O<sub>2</sub> evolution and CO<sub>2</sub> consumption.  $P_N$  increased with higher NaHCO<sub>3</sub> concentrations until reaching the saturation point at 20 mM (Fig. 1C). The high bicarbonate saturation point indicated that our method needed very rich carbon dioxide source. Particularly, in submerged plants,  $P_N$  may be limited by a low availability of dissolved inorganic carbon (Maberly and Spence 1983, Adamec 1997). The interesting finding was that although there was no significant difference between O<sub>2</sub> evolution and CO<sub>2</sub> consumption, the O<sub>2</sub> evolution values were always higher than those of the CO<sub>2</sub> consumption values under the different pH and NaHCO<sub>3</sub> concentrations. This result is important if we want to explore the PQ of *R. mucronata* leaves under aqueous conditions. To be useful, PQ should be determined using the net rate of O<sub>2</sub> involved per CO<sub>2</sub> fixed simultaneously and can be described as:

$$PQ = \frac{O_2 \text{ evolution}}{CO_2 \text{ consumption}}$$

Stoichiometrically, the PQ value equal to 1.00 assumes a hexose production with ammonium as the N source (Rosenberg *et al.* 1995). If this simple photosynthesis physiology was replaced by an ecological summation of protoplasm production, including carbohydrates, proteins, lipids, and nucleic acids, then the theoretical PQ would be higher (Williams and Robertson 1991). Theoretical PQ values typically range from 1.0 to 1.3 (Rosenberg *et al.* 1995). The PQ values of *R. mucronata* leaves under the different pH and NaHCO<sub>3</sub> concentrations ranged from 1.04–1.28 with no significant

difference among them (Fig. 1D). Purely based on stoichiometric and theoretical considerations of the PQ values, results similar or higher than 1.0 would be expected. PQ of 1.0 infers that the sole product of photosynthesis is carbohydrate, while PQ>1.0 indicates that more reduced compounds are produced, such as fats and proteins (Chisholm 1998). Our results also suggested that the simultaneous measurement of O<sub>2</sub> evolution and CO<sub>2</sub> consumption by using the polarographic sensor of Clark oxygen electrode and the 'pCO<sub>2</sub> mini' optode sensor provided simple, stable, and precise measurements of net PQ under aqueous conditions.

The net PQ values in all measurements was never less than 1.0 and it confirmed that photorespiration did not occur under aqueous conditions. A possible explanation for a PQ lesser than 1.0 would be photorespiration (glycolate production) as a result of oxygenase activity of Rubisco at high ambient oxygen concentrations (Rosenberg *et al.* 1995). Photorespiration that decreases PQ occurs when Rubisco, which principally functions as carboxylase, is substituted by its oxygenase function (Taddei *et al.* 2008). In terrestrial C<sub>3</sub> plants, photorespiratory consumption of O<sub>2</sub> can account for 25% of Rubisco activity (Falkowski and Raven 1997). Conversely, photorespiration is assumed to be of minor importance to aquatic plants compared with terrestrial C<sub>3</sub> plants (Laws *et al.* 2000), because submerged environmental conditions, such as fairly constant oxygen and total inorganic carbon concentrations, does not favour photorespiration (Rosenberg *et al.* 1995).

In order to characterise the functioning of photosynthetic apparatus of *R. mucronata* in air and under aqueous conditions, the light curves of  $P_N$  for similarly paired leaves were estimated. In Fig. 2, at low light levels [PAR < 500 μmol(photon) m<sup>-2</sup> s<sup>-1</sup>], the photosynthetic rate of O<sub>2</sub> evolution and CO<sub>2</sub> consumption under aqueous conditions was lower than the photosynthetic CO<sub>2</sub> exchange in air. This result is likely to be related to the reduction of low light utilisation while the leaf slices were rotated under aqueous conditions. Another possible explanation was that our method worked well under light saturation compared to light limitation. Therefore, we needed to improve the simultaneous measurements of photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption under aqueous conditions under low light conditions.

The light saturation points for all  $P_N$  measurements (CO<sub>2</sub> exchange in air, O<sub>2</sub> evolution under aqueous condition, and CO<sub>2</sub> consumption under aqueous condition) were similar at PAR levels around 500–1,000 μmol(photon) m<sup>-2</sup> s<sup>-1</sup>. The  $P_{max}$ , which demonstrates the potential photosynthetic capacity of *R. mucronata* leaves (Ulqodry *et al.* 2014), was also determined. All experiments produced comparable results with similar  $P_{max}$  values of 13.37, 13.11, and 12.31 μmol m<sup>-2</sup> s<sup>-1</sup> for CO<sub>2</sub> exchange in air, O<sub>2</sub> evolution under aqueous conditions, and CO<sub>2</sub> consumption under aqueous conditions, respectively. In comparison with gas



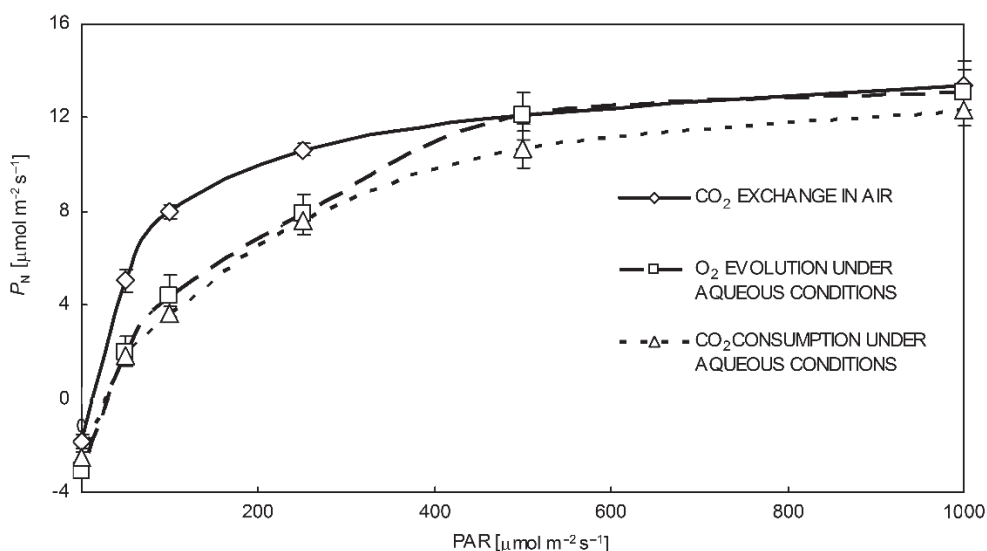


Fig. 2. Response of net photosynthetic rate ( $P_N$ ) to increasing PAR in the *R. mucronata* leaves. Measurements in air were made with a portable open-flow gas-exchange system, LI-6400 (CO<sub>2</sub> exchange in air) and measurements under aqueous conditions were made simultaneously with an aqueous-phase O<sub>2</sub> electrode (O<sub>2</sub> evolution) and 'pCO<sub>2</sub> mini' optode system (CO<sub>2</sub> consumption). Temperature was 25°C for all measurements. Values are means  $\pm$  SD.

exchange, the maximum photosynthetic rate in photosynthetic O<sub>2</sub> evolution and CO<sub>2</sub> consumption under aqueous condition was achieved under very high carbon dioxide concentration. The  $P_{max}$  value and daily period of irradiance, when plants were in the water and air,

would be useful as an indicator of primary production (Zimmerman *et al.* 1994). The similar  $P_{max}$  values suggested that all treatments resulted in a high capacity to adjust the photosynthetic apparatus under light saturation conditions.

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