# Antioxidant Activity of Extracts of Halodule pinifolia Seagrass from Solvents with Different Polarities

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# Antioxidant Activity of Extracts of *Halodule pinifolia*Seagrass from Solvents with Different Polarities

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#### **ABSTRACT**

The purpose of this study was to analyze phytochemistry contents and antioxidant activity of extracts from seagrass of *Halodule pinifolia* from solvents with different polarities. Parameters of research were phytochemical content, DPPH scavenging activity and reducing power. The result showed content of phytochemical compounds of ethanol extract seagrass were flavonoids, tannins, saponins, steriods and triterpenoids. The use of ethyl acetate solvent showed phytochemical compounds were flavonoids, steroids and triterpenoids. For n-hexane solvent showed phytochemical compounds were steroids and triterpenoids. The highest of antioxidant activity with DPPH method (IC<sub>50</sub>) of *H. pinifolia* was 18.7 ppm with ethanol extract. The highest of reducing power of *H. pinifolia* was 1.749.

Keywords: Seagrass, Halodule pinifolia, antioxidant.

#### INTRODUCTION

Seag 14 ses are flowering plants (angiosperms) which grow in marine, fully saline environments. Seagrasses are a rich source of structurally novel and biologically active metabolites which they produce in order to sustain the extreme environmental conditions prevailing under sea<sup>1</sup>

Seagrasses produce antioxidant compounds that inhibits the oxidation of other molecules and there are many reports describing antioxidant activities<sup>2-5</sup>, antifungal<sup>6</sup>, antiviral<sup>7</sup>, anti-inflammatory<sup>8</sup>, antidiabetic<sup>9</sup> and antibacterial<sup>10-12</sup>.

However reports on the phytochemical constituents of seagrasses of their bioactive activity of Indonesian sea are limited with the exception of few studies in this research 13,14, we reported that antioxidant activity of extracts of *Halodule pinifolia* seagrass from solvents with different polarities.

#### **MATERIALS AND METHOD**

#### Preparation of Seagrass extract

Extraction of *Halodule pinifolia* seagrass by stratified maceration method using n-hexane, ethyl acetate and ethanol . Sea grass powder were soaked in 2 L with solvent (1:4 w/v), and kept for 2

x 24 h in a shaker. The solution is filtered using the number 42 Whatman filter paper to obtain the filtrate. The filtrate is dried using a freeze dryer to remove the solvent that may remain in the extract

#### Phytochemical Screening of Halodule pinifolia

Test of flavonoids, alkaloids, saponin, steroids, triterpenoids were determined by Harborne method<sup>15</sup>.

#### 10 DPPH radical scavenging activity

DPPH radical scavenging activity was measured based on methods described in Hanani et al.<sup>16</sup>.

#### 16 Reducing power

Reducing power was determined by Oyaiza method<sup>17</sup>.

#### **RESULT AND DISCUSSION**

#### The phytochemical screening

As seen as Table 1 showed content of phytochemical compounds of extract of seagrass were flavonoids, alkaloids, tannins, saponins, steroids and triterpenoids.

The result showed content of phytochemical compounds of ethanol extract seagrass were flavonoids, tannins, saponins, steriods and triterpenoids. The use of ethyl acetate solvent showed the phytochemical compounds were flavonoids, steroids and triterpenoids. For n-hexane solvent showed phytochemical compounds were steroids and triterpenoids.

### DPPH radical scavenging activity

Method of DPPH radical scavenging activity is very popular for the research of natural

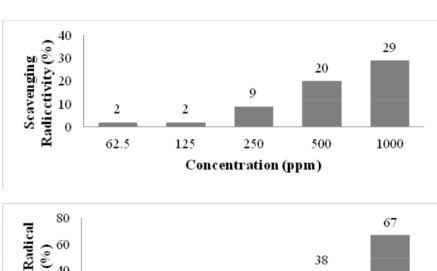
Table 1: Phytochemical compound of extract of *H. pinifolia* seagrass

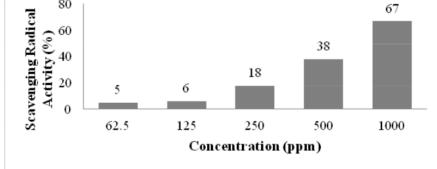
Sample	Parameter		Result
<i>n-</i> hexane	Flavonoids		13 Negative
	Alkaloids	Wegner	Negative
		Mayer	Negative
		Dragendorf	Negative
	Tannins		Negative
	Saponins		Negative
	Steroids		Positive
	Triterpenoids		75 itive
Ethyl acetate	Flavonoids		Positive
	Alkaloids	Wegner	Negative
		Mayer	Negative
		Dragendorf	Negative
	Tannins		Positive
	Saponins		Negative
	Steroids		Positive
	Triterpenoids		gositive
Ethanol	Flavonoids		Positive
	Alkaloids	Wegner	Negative
		Mayer	Negative
		Dragendorf	Negative
	Tannins		Positive
	Saponins		Positive
	Steroids		Positive
	Triterpenoids		Positive

antioxidants <sup>18</sup>. The extraction with solvents of increasing polarity involves of separating compounds of a plant according to their degree of a ubility. DPPH radical scavenging activity of hexane, ethyl acetate and ethanol extracts arisined of the Halodule pinifolia were shown in Figure 1. The maximum DPPH radical scavenging activity was

recorded in ethanol extracts followed by ethyl acetate and n-hexane.

The IC 50 of extract was 18.7 ppm for ethanol extract, 696.2 ppm for ethyl acetate extract and 2,378.2 ppm for n-hexane extract. The IC 50 value for vitamin C was 7.7 ppm (Figure 2). The results





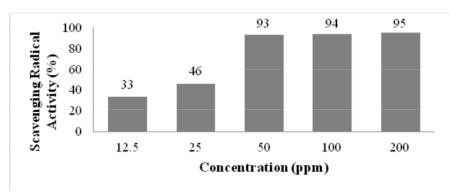


Fig. 1: Scavenging radical DPPH activity extract of *H. pinifolia* Seagrass (A= n-hexane, B=ethyl acetic and C= ethanol)

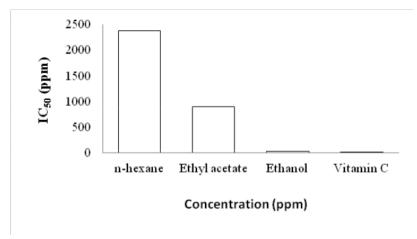


Fig. 2: IC<sub>50</sub> from extract of *H. pinifolia* Seagrass

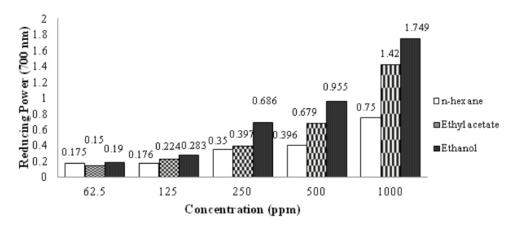


Fig. 3; Reduction power of extract of H. pinifolia Seagrass

indicate that the antioxidant activity 4 the methanol extract of *H. pinifolia* seagrass is higher than that of ethyl acetate and n-hexane extracts. Antioxidant activity of extract *Halodule pinifolia* could be due to their phytochecological compounds. The phytochemical compounds present in the extract, which are responsible for this activity. The phytochemical tests indicated the presence of flavonoids, tannins, saponins, steriods and triterpenoids in the crude methanolic extract.

#### 6 Reducing power

Reducing power of extract of *H. pinifolia* depicted in Figure 2. In 6 asing of concentration of *H. pinifolia* indicates an increase in reducing power

The reducing power is considered as a significant indicator of potential antioxidant activity of compound or sample. A potential antioxidant will reduce the ferric ion to the ferrous ion. Reducing power of extract of *H. pinifolia* is probably due to the presence of phytochemical compounds that can serve as an electron donor.

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